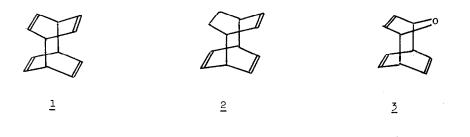
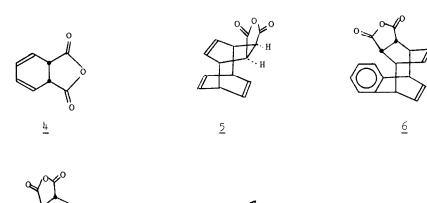
PHOTOCYCLOADDITION OF <u>CIS</u>-1,2-DIHYDROPHTHALIC ANHYDRIDE TO BENZENE AND ARENES N. C. Yang,<sup>\*1</sup> Charles V. Neywick, and Kasturi Srinivasachar

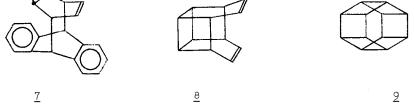
Department of Chemistry, University of Chicago, Chicago, Illinois 60637

(Received in USA 27 August 1975; received in UK for publication 24 October 1975)

Tricyclo[4.2.2.2<sup>2,5</sup>]dodeca-3,7,9,11-tetraene (1) and tricyclo[4.2.2.2<sup>2,5</sup>]dodeca-3,7,9triene  $(\underline{2})$  contain a number of isolated  $\pi$ -systems in spatial proximity and thus are expected to exhibit interesting chemical and physical properties. Although the photocycloadditions of cyclic 1,3-dienes to aromatic hydrocarbons yield several novel products, 2-5 benzene itself does not react photochemically with 1,3-cyclohexadiene to give the expected adduct (2). The major volatile products of this reaction are isomeric 1,3,5-hexatrienes under a variety of experimental conditions,<sup>6</sup> in direct contrast to the photocycloaddition of furan to benzene which gives the  $4\pi_s+4\pi_s$  adduct (3), recently reported by Cantrell and by Berridge, Bryce-Smith and Gilbert.<sup>4,5</sup> Apparently, benzene may transfer its excitation energy to 1,3-cyclohexadiene which then undergoes a conrotatory ring-opening to give cis-1,3,5-hexatriene,  $^7$  while such an energy transfer cannot occur with furan. If the ring opening is retarded by appropriate substituents in the cyclohexadiene moiety, the  $4\pi_s+4\pi_s$  photocycloaddition between benzene and the substituted 1,3-cyclohexadiene may proceed to give a substituted  $\underline{2}$  among its products. We wish to report the photocycloaddition of cis-1,2-dihydrophthalic anhydride  $(\frac{1}{2})$  to benzene, naphthalene, and anthracene to give 5-7 respectively. These compounds may be the key intermediates in the syntheses of homo- and heterodimers of benzene including  $\underline{1}$  and other novel cyclic systems, such as 8 and 9.





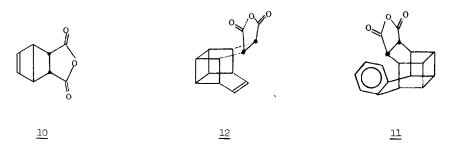


Irradiation of a  $1.67 \times 10^{-2}$  M solution of  $\frac{1}{2}$  in benzene (1 g/400 ml) under nitrogen with a Hanovia 450-w Hg-arc through a Vycor filter yielded a mixture including a copious amount of an insoluble yellow deposit which was not identified. Evaporation of the benzene solution yielded 880 mg of a residue, from which two major products could be isolated by chromatography on silica gel, bicylo[2.2.0]hex-5-ene 2,3-dicarboxylic acid anhydride <u>10</u> isolated as the diacid (145 mg,  $29\%)^8$  and 5/(76 mg, 5%), plus traces of several minor products. The crude yields of 5/2 and 10 may be estimated from the nmr spectrum of the reaction mixture to be 13% and 51% respectively. Recrystallization of 5 from benzene-hexane yielded an analytical sample, mp 149-150° (dec); uv end absorption; ir (KBr) 1785 and 1850 cm<sup>-1</sup>; mass spec (70 ev) parent ion (m/e) 228; anal. found C, 73.53 and H, 5.35; nmr  $\delta(\text{CDCl}_3)$  3.39 (m, 2), 3.50 (m, 2), 3.86 (s, 2), 6.00 (sextet, 2), 6.29 (sextet, 2) and 6.52 ppm (sextet, 2). Spin decoupling experiments substantiated the structure assignment. Photocycloaddition of  $rac{1}{2}$  to naphthalene and anthracene in dichloromethane proceeded smoothly to give  $\underline{6}$  (85%) and  $\underline{7}$  (69%), respectively. The structure of  $\underline{6}$  and  $\underline{7}$  were established by spectroscopic and elemental analyses. The hydrogens  $\alpha$ - to the anhydride function occupy the endo positions as indicated by the lack of coupling with the neighboring bridgehead hydrogens in their nmr spectra [ $\delta$  3 ppm (s)]. The structure of <u>6</u> was further substantiated

by the xanthone sensitized photocyclization to <u>11</u> [66% yield; mp 240-241°; ir (KBr) 1840 and 1745 cm<sup>-1</sup>; uv max (ethanol) 273.5 ( $\epsilon$  128) and 264 nm ( $\epsilon$  177); nmr  $\delta$  (CDCl<sub>3</sub>) 2.62 (s, 2), 2.90-3.15 (m, 4), 3.73 (m, 2), 4.10 (m, 2) and 7.10-7.25 ppm (m, 4); anal. found C, 77.21 and H, 5.01].

Since <u>10</u> is the major product in the irradiation of a very dilute solution of  $\frac{1}{2}$  in benzene, in which virtually all the incident light was absorbed by benzene, energy transfer from excited benzene to  $\frac{1}{2}$  is still the major reaction pathway. The excited singlet energy levels of 1,3cyclohexadiene ( $\lambda_{max}$  264 nm,  $\epsilon$  8,000) and  $\frac{1}{2}$  ( $\lambda_{max}$  264 nm,  $\epsilon$  5,000) are not known. Judging from their uv spectra and our experimental results, they must be quite close to each other as well as to that of benzene such that a minor modification in the structure of 1,3-cyclohexadiene to  $\frac{1}{2}$  enables the  $4\pi_s + 4\pi_s$  addition to occur.

Preliminary investigations indicated that  $\frac{5}{2}$  also undergoes xanthone sensitized photocyclization to  $\underline{12}$ . The conversion of  $\frac{5}{2}$  to other ring systems and the substituent effect in aromatic hydrocarbons on their photocycloaddition to  $\underline{4}$  are currently under investigation in our laboratory.



<u>Acknowledgement</u>. The authors wish to thank the National Institute of General Medical Sciences, grant GM-20329, and the National Science Foundation, grant MPS-22579, for the support of this work. The high resolution nmr spectrometers used were purchased with funds from equipment grants made by the National Science Foundation (GP-33116) and the National Cancer Institute through the University of Chicago Cancer Research Center (CA-14599).

## Footnotes

(1) Guggenheim Fellow, 1974-5.

(2) K. Kraft and G. Koltzenburg, Abst. Internat. Conf. Photochem. (Munich), 439 (1967) [Chem. Abst., 70, 57309 (1969)]; Tetrahedron Letters, 4357 and 4723 (1967).

(3) N. C. Yang and J. Libman, Tetrahedron Letters, 1409 (1973), and references therein.

(4) T. S. Cantrell, Tetrahedron Letters, 3959 (1974).

(5) J. C. Berridge, D. Bryce-Smith, A. Gilbert, and T. S. Cantrell, J. Chem. Soc. Chem. Commun., 611 (1975); J. C. Berridge, D. Bryce-Smith, and A. Gilbert, <u>ibid.</u>, 964 (1974); Tetrahedron Letters, 2325 (1975).

(6) N. C. Yang and J. Libman, unpublished results.

(7) R. B. Woodward and R. Hoffmann, "Conservation of Orbital Symmetry," Academic Press, New York, N. Y., 1971, p. 45.

(8) E. E. van Tamelen and S. P. Pappas, J. Amer. Chem. Soc., 85, 3297 (1963).